NCLEX Notes Week 4: Fluid, Electrolytes, and ABGs

Electrolytes

- a substance that dissolves into a solution and ionizes (some of its molecules split or dissociate into electrically charged atoms or ions)
 - Examples of Electrolytes: Potassium, Sodium, Chloride, Magnesium

Homeostasis

- Stability for internal conditions
 - The number of cations and anions must be the same for homeostasis to exist!

Extracellular Fluid

- Fluid outside of the cells
 - Examples of Extracellular Fluid: blood, lymph, and plasma,

Edema

- excess fluid accumulation in the interstitial space
 - **Localized edema:** comes as the result of trauma, from an accident or surgery, burns, or inflammatory processes
 - **Generalized Edema/ Anasarca:** excessive fluid in the interstitial space caused by conditions like cardiac or renal failure

Types of IV Solutions

- Isotonic
 - \circ $\;$ When solutions on both sides are equal in concentration
- Hypotonic
 - $\circ~$ A hypotonic solution has less salt and more water
- Hypertonic
 - A hypertonic solution has more salt and less water

Insensible Loss

- Water lost through the skin
 - \circ $\,$ Examples of insensible Loss: Sweat, through breathing

Hypovolemia

- Most common form of dehydration
- Water and dissolved electrolytes are lost in equal proportions

Causes fo fluid volume deficits

- Inadequate fluid intake
- Fluid shift between compartments
- Conditions that increase fluid loss like ketoacidosis prolonged fevers, and diabetes insipidus
- Chronic illness
- Kidney Disease
- Hypervolemia
- Excessive fluid in the extracellular compartment

1 otussium (3.3 to 3)		
Hypokalemia	Hyperkalemia	
 Potassium is lower than 3.5 Can be caused by medications like diuretics, vomiting/diarrhea, nasogastric suctioning, and kidney disfunction 	 Potassium is higher than 5 Can be caused by excessive potassium intake, kidney problems, potassium retaining diuretics, and tissue damage 	
A SIC WALT	MURDER	
 Alkalosis Shallow respirations Irritability Confusion and drowsiness Weakness and fatigue Arrhythmias Lethargy Thready pulse 	 Muscle cramps Urine abnormalities Respiratory Distress Decreased cardiac contractility EKG changes Reflexes 	

Potassium (3.5 to 5)

Sodium (135-145)

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Hyponatremia	Hypernatremia
 Sodium is lower than 135 Can be caused by vomiting/diarrhea, diuretics, kidney disease, excessive diaphoresis 	 Sodium is higher than 145 Can be caused by kidney disease, increased sodium intake, and decreased water intake

SALTLOSS	FRIED
 Stupor Anorexia Lethargy Tendon reflexes decreased Limp muscles Orthostatic hypotension Seizures/headaches Stomach cramping 	 Fever Restlessness Increased fluid retention Edema Decreased urinary output

Calcium (9 - 10.5)

Hypocalcemia	Hypercalcemia
 Calcium is lower than 9 Can be caused by lactose intolerance, end-stage kidney disease, inadequate calcium intake, and Crohn's disease 	 Calcium is higher than 10.5 Can be caused by decreased calcium excretion, hyperthyroidism, and excessive vitamin intake
• Convulsions Arrhythmias Tetany Spasm or stridor	 Bone pain Arrhythmias Cardiac Arrest Kidney Stones Muscle Weakness Excessive Urination

Magnesium (1.3 - 2.1)

Hypomagnesemia	Hypermagnesemia
 Magneisum is lower than 1.3 Can be caused by vomiting/diarrhea, malnutrition, Crohn's disease, and alcoholism 	 Magnesium is higher than 2.1 Can be caused by increased magnesium intake. Decreased renal excretion of magnesium
ConfusionIncreased Deep tendon reflexes	FlushingMuscle weakness

Phosphate(3.0 - 4.5)

Hypophosphatemia	Hyperphosphatemia
 Phosphate is lower than 3.0 Can be caused by intestinal malabsorption, respiratory alkalosis 	 Phosphate is higher than 4.5 Can be caused by renal failure and long term laxative use
 Irritability Confusion Convulsions Respiratory failure 	• Similar to Hypocalcemia

Acid-Base Imbalance

INTERPRETATION OF ABG			
ACID BASE	Ph	PaCO ₂	HCO ₃
NORMAL	7.35	35-45	22-26
RESP. ACIDOSIS	\bigcup		NORMAL
RESP. ALKALOSIS		\bigcup	NORMAL
METABOLIC ACIDOSIS	\bigcup	NORMAL	\bigcup
METABOLIC ALKALOSIS		NORMAL	

Acid-Base Mnemonic—ROME		
Respiratory		
O pposite		
 Alkalosis 	↑ pH	↓ PaCO2
• Acidosis	↓pH	↑ PaCO2
Metabolic		
Equal		
• Acidosis	↓pH	↓ HCO ₃ ⁻
 Alkalosis 	↑pH	↑ HCO ₃ [–]

Respiratory Acidosis

• Can be caused by asthma, bronchitis, COPD, or pulmonary edema

Clinical Manifestations of Respiratory Acidosis

- Tachycardia
- V-Fib
- Anxiety
- Confusion
- Shallow breathing
- Increased respiratory rate
- pale or cyanotic skin
- increased blood pressure

Respiratory Alkalosis

• Can be Caused by fever, Hyperventilation, Hypoxia, or hysteria

Clinical Manifestations of Respiratory Alkalosis

- Tachypnea
- numbness and tingling
- tinnitus
- tachycardia

- dysrhythmias
- Rapid Deep Respirations
- Changes in the level of consciousness

Metabolic Acidosis

• Diabetic Ketoacidosis, malnutrition renal insufficiency

Clinical Manifestations of Metabolic Acidosis

- Dysrhythmias
- Bradycardia
- Hypotension
- warm, pink, dry skin tachypnea
- headache
- drowsiness
- Kussmaul Respirations (try to blow off more CO2)

Metabolic Alkalosis

• Diuretics, excessive vomiting, or gastrointestinal suctioning

Clinical Manifestations of Metabolic Alkalosis

- Tachycardia
- dysrhythmia
- hyper-reflexes
- drowsiness
- anorexia
- n/v
- muscle cramps
- confusion
- convulsions/seizures ineffective breathing patterns